

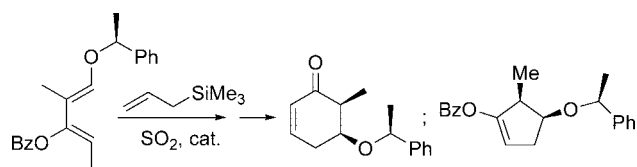
## Synthesis of Optically Active 5-Alkoxy-6-methylcyclohex-2-en-1-ones and 4-Alkoxy-5-methylcyclopent-1-enyl Benzoate

Māris Turks<sup>†</sup> and Pierre Vogel\*

Laboratory of Glycochemistry and Asymmetric Synthesis  
(LGSA), Swiss Federal Institute of Technology (EPFL),  
Batochime, CH-1015 Lausanne, Switzerland

pierre.vogel@epfl.ch

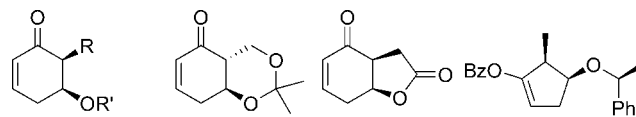
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The reaction of (–)-(1*E*,3*Z*)-2-methyl-1-((1*S*)-1-phenylethoxy)penta-1,3-dien-3-yl benzoate with allyltrimethylsilane in SO<sub>2</sub> and in the presence of a catalytic amount of Tf<sub>2</sub>NTMS gives a silyl sulfinate intermediate that furnishes (–)-(6*Z*,1'*S*,4*S*,5*S*)-5-methyl-4-(1'-phenylethoxy)octa-1,6-dien-6-yl benzoate after acidic workup. The latter undergoes ring-closing metathesis producing (–)-(2*S*,3*S*)-2-methyl-3-((1*S*)-1-phenylethoxy)cyclopent-5-en-1-yl benzoate. It has been converted also into the corresponding trimethylsilyl enol ether. After oxidation, an enone is obtained that undergoes ring-closing metathesis giving (–)-(5*S*,6*S*)-6-methyl-5-((1*S*)-1-phenylethoxy)cyclohex-2-en-1-one.

Enantiomerically enriched cyclohex-2-enone derivatives are extremely useful building blocks for the preparation of biologically active compounds, including natural products.<sup>1–8</sup> Among the various types of cyclohex-2-enone derivatives, 5-hydroxy- and 5-alkoxycyclohex-2-enones represent the most attractive and versatile chiral building blocks,<sup>9–15</sup> and several approaches for

their synthesis have been proposed.<sup>16,17</sup> The two 5-siloxycyclohex-2-enones **1** and **2**<sup>18</sup> and bicyclic compounds **3**<sup>19</sup> and **4**<sup>20</sup> are the only enantiomerically enriched 6-alkyl-substituted 5-hydroxycyclohex-2-enone derivatives reported thus far.



- 1** R = Me, R' = TBS  
**2** R = CH<sub>2</sub>Ph, R' = TBS  
**5** R = Me, R' = (1*S*)-1-phenylethyl

The synthesis of 5-hydroxy-, 5-alkoxy-, and 5-siloxycyclohex-2-enones is a challenge because these compounds are prone to β-eliminations generating the corresponding phenols, under basic or acidic conditions. We present here an original approach to the synthesis of enantiomerically enriched 6-alkyl-5-alkoxycyclohex-2-enones. The method is illustrated for the preparation of derivative **5**. We disclose also the synthesis of the enantiomerically enriched cyclopentanone enol benzoate **6**, a yet unknown compound. Apart from their potential as synthetic intermediates, both **5** and **6** are of interest for the perfume industry.<sup>21–32</sup>

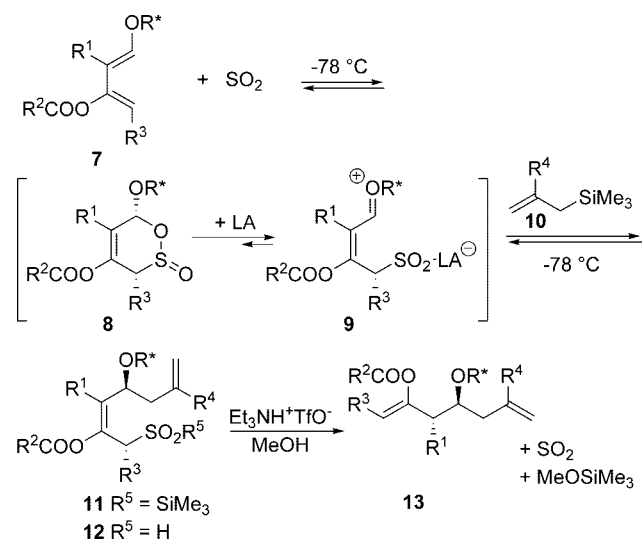
In 2004, we reported that the reactions of 1-alkoxy-1,3-dien-3-ol esters (**7**) with sulfur dioxide and allylsilanes (**10**) generate enol esters of type **13** with high stereoselectivity. The reactions imply equilibrium of dienes **7** + SO<sub>2</sub> with corresponding

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<sup>†</sup> Present Address: Faculty of Material Science and Applied Chemistry, Riga Technical University, Riga LV-1048, Latvia.

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## SCHEME 1

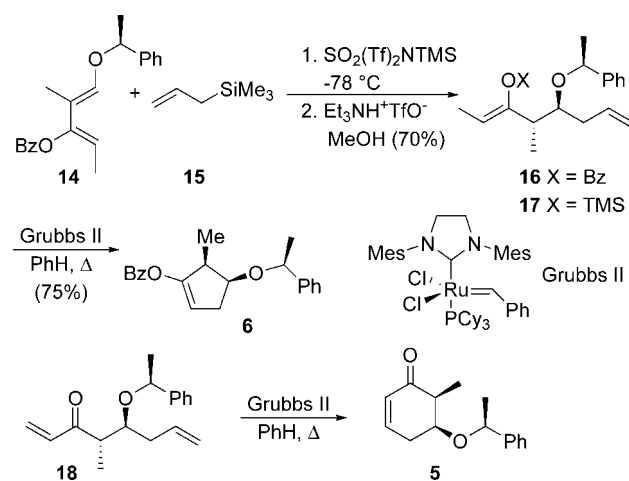


unstable sultines **8** that, in the presence of a protic or Lewis acid catalyst, are isomerized into zwitterionic species **9** that are allylated by the allylsilanes **10** producing silyl sulfonates **11**. After acidic workup, the corresponding  $\beta,\gamma$ -unsaturated sulfonic acids **12** undergo stereoselective retro-ene eliminations of  $\text{SO}_2$  with formation of **13** (Scheme 1).<sup>33</sup>

In this process, sulfur dioxide umpolung permits the Umpolung of the electron-rich dienes **7** generating oxallyl cationic intermediates (**9**) that are quenched by electron-rich alkenes such as allylsilanes.<sup>34,35</sup> Compounds **13** undergo ring-closing metathesis<sup>36–38</sup> catalyzed by the second-generation Grubbs' catalyst.<sup>39–42</sup> An example is given with enol benzoate **16** (ee 97%) obtained by reaction of (–)-(1*E*,3*Z*)-2-methyl-1-((1*S*)-1-phenylethoxy)-penta-1,3-dien-3-yl benzoate (**14**, ee 97%) with  $\text{SO}_2$  and allyltrimethylsilane (**15**) in the presence of a catalytic amount of  $(\text{CF}_3\text{SO}_2)_2\text{NSiMe}_3$  at  $-78^\circ\text{C}$ .<sup>43</sup> Thus, in the presence of 0.1 equiv of Grubbs' II catalyst (dry benzene,  $80^\circ\text{C}$ , 30 min), cyclopentenyl benzoate **6** was obtained in 75% yield. To our knowledge, this is the first example of ring-closing metathesis involving an enol ester and an alkene (Scheme 2), although several examples of ring-closing alkene metatheses involving enol ethers or enoxysilanes with alkenes have been presented.<sup>44,45</sup>

Enol benzoate **16** was converted into its silyl enol ether **17** by reaction with  $\text{MeLi}\cdot\text{LiBr}$  in  $\text{Et}_2\text{O}$  at  $-78^\circ\text{C}$  to generate the corresponding ethyl ketone that was not purified but used directly in the silylation step with  $\text{Me}_3\text{SiOTf}/\text{Et}_3\text{N}$ . Enoxysilane

## SCHEME 2



**17** was obtained in 82% yield (based on **16**) and was then submitted to oxidation with an excess of 2-iodoxybenzoic acid (IBX) and 4-methoxypyridine *N*-oxide (MPO) in DMSO ( $25^\circ\text{C}$ , 16 h)<sup>46,47</sup> to produce enone **18** in 64% yield. In the presence of 10 mol % of Grubbs' II catalyst in dry benzene ( $80^\circ\text{C}$ , 30 min), ring-closing alkene metathesis furnished cyclohexenone **5** in 85% yield. Because of the neutral conditions used for the latter reaction, heating of **5** did not induce  $\beta$ -elimination of 1-phenylethanol.

We believe that the chemistry disclosed here can be applied to the synthesis to further enantiomerically enriched 5-alkoxy-6-alkylcyclohex-2-enone derivatives starting from dienes of type **7** ( $\text{R}^1 \neq \text{Me}$ ). As both enantiomeric forms of the chiral auxiliary (1-phenylethanol) are commercially available, the cyclohexenones can be obtained in both their enantiomeric forms.

## Experimental Section

**Preparation of (–)-(2*S*,3*S*)-2-Methyl-3-((1*S*)-1-phenylethoxy)-cyclopent-5-en-1-yl Benzoate (**6**).** To a solution of Grubbs' second-generation catalyst (31 mg, 0.036 mmol, 0.1 equiv) in dry benzene (70 mL) under argon atmosphere was added a solution of **16** (0.133 g, 0.36 mmol, 1 equiv).<sup>43</sup> The resulting solution was heated under reflux for 0.5 h. Solvent was evaporated in vacuo and the residue purified by FC (PE/EtOAc = 95:5): 88 mg (75%), colorless oil;  $[\alpha]_{\text{D}}^{25} = -176$  ( $c = 1.40$ ,  $\text{CHCl}_3$ );  $^1\text{H NMR}$  ( $\text{CD}_2\text{Cl}_2$ , 400 MHz) 8.07 (d, 2H,  $J = 8.0$  Hz), 7.60 (t, 1H,  $J = 7.0$ ), 7.47 (t, 2H,  $J = 7.7$ ), 7.36–7.25 (m, 5H), 5.44 (ddd, 1H,  $J = 2.5$ , 2, 1.1), 4.46 (q, 1H,  $J = 6.4$ ), 4.13 (dt, 1H,  $J = 7.4$ , 6.8), 3.00 (dq, 1H,  $J = 7.4$ , 6.8, 1.5), 2.42 (ddd, 1H,  $J = 16.0$ , 7.4, 3.2), 2.34 (dddd, 1H,  $J = 16.0$ , 6.8, 2.5), 1.43 (d, 3H,  $J = 6.4$ ), 1.18 (d, 3H,  $J = 7.0$ );  $^{13}\text{C NMR}$  ( $\text{CD}_2\text{Cl}_2$ , 100.6 MHz) 164.2, 152.8, 144.5, 133.4, 129.8, 128.6, 128.4, 127.4, 126.4, 126.3, 109.3, 76.9, 76.1, 40.5, 34.4, 24.0, 11.2; MALDI-HRMS calcd. for  $\text{C}_{21}\text{H}_{22}\text{O}_3\text{Na}^+$  345.1467, found 345.1480.

**(+)-(4*S*,5*S*)-4-Methyl-5-((1*S*)-1-phenylethoxy)octa-1,7-dien-3-one (**18**).** A solution of **16** (3.32 g, 9.1 mmol, 1 equiv) in  $\text{Et}_2\text{O}$  (50 mL) was added at  $-78^\circ\text{C}$  to a solution of  $\text{MeLi}\cdot\text{LiBr}$  (2.2 M, 10.3 mL, 22.7 mmol, 2.5 equiv) in  $\text{Et}_2\text{O}$  (24 mL). The reaction mixture was stirred at  $-78^\circ\text{C}$  overnight and afterward poured slowly into a saturated aqueous solution of  $\text{NH}_4\text{Cl}$  (200 mL). The aqueous phase was extracted with  $\text{Et}_2\text{O}$  ( $3 \times 50$  mL), washed with brine, dried ( $\text{Na}_2\text{SO}_4$ ), and evaporated in vacuo. The resulting ethyl

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ketone (2.23 g, ~94%) was dried under reduced pressure (0.06 Torr, 12 h) and used without further purification. To a solution of the ethyl ketone (2.23 g, 8.56 mmol, 1 equiv) in CH<sub>2</sub>Cl<sub>2</sub> (40 mL) at -20 °C was added NEt<sub>3</sub> (3.58 mL, 25.7 mmol, 3 equiv) followed by Me<sub>3</sub>SiOSO<sub>2</sub>CF<sub>3</sub> (2.01 mL, 11.1 mmol, 1.3 equiv). The reaction mixture was allowed to reach 20 °C in 2 h. Cold pentane (300 mL (-50 °C)) was added, and the resulting suspension was filtered through a small pad of silica gel (suspended in pentane containing 2% of NEt<sub>3</sub>). The organic phase was sequentially washed with 15% aqueous solution of citric acid (3 × 70 mL), saturated aqueous solution of NaHCO<sub>3</sub> (3 × 50 mL), and brine (2 × 50 mL), dried (Na<sub>2</sub>SO<sub>4</sub>), and evaporated in vacuo. The resulting oil (**17**) was dried under reduced pressure (0.06 Torr, 12 h): yield 2.49 g (82% over two steps from **16**), colorless oil. Analytical data for (2*S*,3*S*)-(1-ethylidene-2-methyl-3-((1*S*)-1-phenylethoxy)hex-5-enyloxy)trimethylsilane (**17**): <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz) 7.36–7.25 (m, 5H), 5.65 (ddt, 1H, *J* = 17.3, 10.2, 7.0), 4.96 (dm, 1H, *J* = 17.3), 4.92 (dm, 1H, *J* = 10.2), 4.62 (q, 1H, *J* = 6.4), 4.52 (q, 1H, *J* = 6.4), 3.53 (dt, 1H, *J* = 7.0, 3.8), 2.52 (dq, 1H, *J* = 7.0, *J* = 6.4), 2.18–2.03 (m, 2H), 1.53 (dd, 3H, <sup>3</sup>*J* = 7.0, 1.3), 1.42 (d, 3H, *J* = 6.4), 1.02 (d, 3H, *J* = 7.0), 0.19 (s, 9H). To a solution of IBX (0.252 g, 0.90 mmol, 3 equiv) in DMSO (1.0 mL) was added MPO·H<sub>2</sub>O (0.112 g, 0.90 mmol, 3 equiv) at 25 °C. The resulting mixture was allowed to stay for 30 min. Then a solution of **17** (0.10 g, 0.30 mmol, 1 equiv) in DMSO (0.5 mL) was added, and the resulting mixture was stirred at 25 °C for 16 h. A saturated aqueous solution of NaHCO<sub>3</sub> (10 mL) was added, and the product was isolated by extraction with Et<sub>2</sub>O (3 × 7 mL). The collected organic layers were washed with brine (3 × 6 mL), dried (Na<sub>2</sub>SO<sub>4</sub>), filtered, and evaporated. The residue was purified by FC (PE/EtOAc = 95:5) yielding 50 mg (64% from **17**): colorless oil; [α]<sub>D</sub><sup>25</sup> = +73 (*c* = 0.45, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz) 7.35–7.24 (m, 5H), 6.44 (dd, 1H, *J* = 17.2, 10.5), 6.25 (dd, 1H, *J* = 17.2,

1.2), 5.78 (dd, 1H, *J* = 10.5, 1.2), 5.74 (ddt, 1H, *J* = 17.2, 10.4, 7.4), 4.99 (dm, 1H, *J* = 10.4), 4.96 (dm, 1H, *J* = 17.2), 4.43 (q, 1H, *J* = 6.2), 3.72 (dt, 1H, *J* = 8.0, 4.9), 3.21 (dq, 1H, *J* = 8.0, 6.8), 2.13 - 2.06 (m, 2H), 1.32 (d, 3H, *J* = 6.2), 1.03 (d, 3H, *J* = 6.8); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100.6 MHz) 203.5, 144.2, 136.5, 134.1, 128.3, 128.1, 127.5, 126.6, 117.3, 78.5, 46.2, 35.5, 23.6, 12.2; ESI-HRMS calcd for [C<sub>17</sub>H<sub>22</sub>O<sub>2</sub> + H]<sup>+</sup> 259.1698, found 259.1692.

(-)-(5*S*,6*S*)-6-Methyl-5-((1*S*)-1-phenylethoxy)cyclohex-2-en-1-one (**5**). To a solution of Grubb's second-generation catalyst (98 mg, 0.116 mmol, 0.1 equiv) in dry benzene (240 mL) under argon atmosphere was added a solution of **18** (0.30 g, 1.16 mmol, 1 equiv). The resulting solution was heated under reflux for 0.5 h. Solvent was evaporated in vacuo, and the residue was purified by FC (PE/EtOAc = 9:1): yield 0.227 mg (85%); colorless oil; [α]<sub>D</sub><sup>25</sup> = -130 (*c* = 0.55, CHCl<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>, 400 MHz) 7.37–7.25 (m, 5H), 6.69 (dt, 1H, *J* = 10.2, 3.8), 6.25 (dt, 1H, *J* = 10.2, 1.9), 4.49 (q, 1H, *J* = 7.0), 3.85 (dt, 1H, *J* = 7.0, 4.5), 2.76 (dq, 1H, *J* = 7.0, 3.8), 2.39 (dddd, 1H, *J* = 19.0, 6.0, 3.2, 1.9), 2.34 (dddd, 1H, *J* = 19.0, 5.1, 1.9, 1.9), 1.42 (d, 3H, *J* = 6.4), 1.23 (d, 3H, *J* = 7.0); <sup>13</sup>C NMR (CDCl<sub>3</sub>, 100.6 MHz) 198.1, 146.1, 144.1, 129.1, 128.5, 127.6, 126.2, 76.4, 75.1, 45.9, 30.0, 24.5, 10.5; MALDI-HRMS calcd for C<sub>15</sub>H<sub>18</sub>O<sub>2</sub>Na<sup>+</sup> 253.1204, found 253.1209.

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**Supporting Information Available:** Analytical details for products **6**, **5**, **17**, and **18**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

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